

# Radish plants after 5 days



COTTON FIBERS



MARC



POPLAR



CONIFER



SOIL



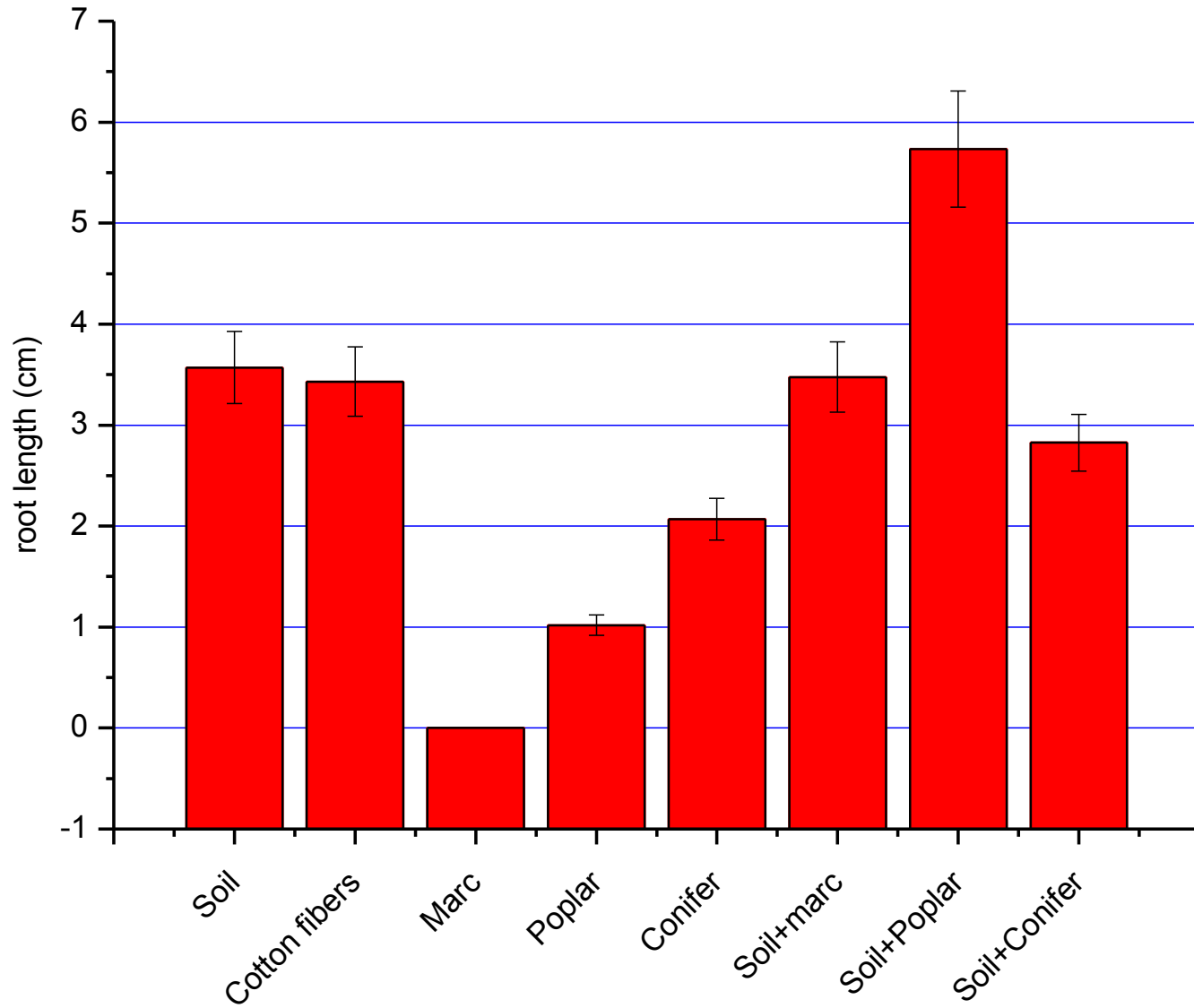
SOIL+MARC



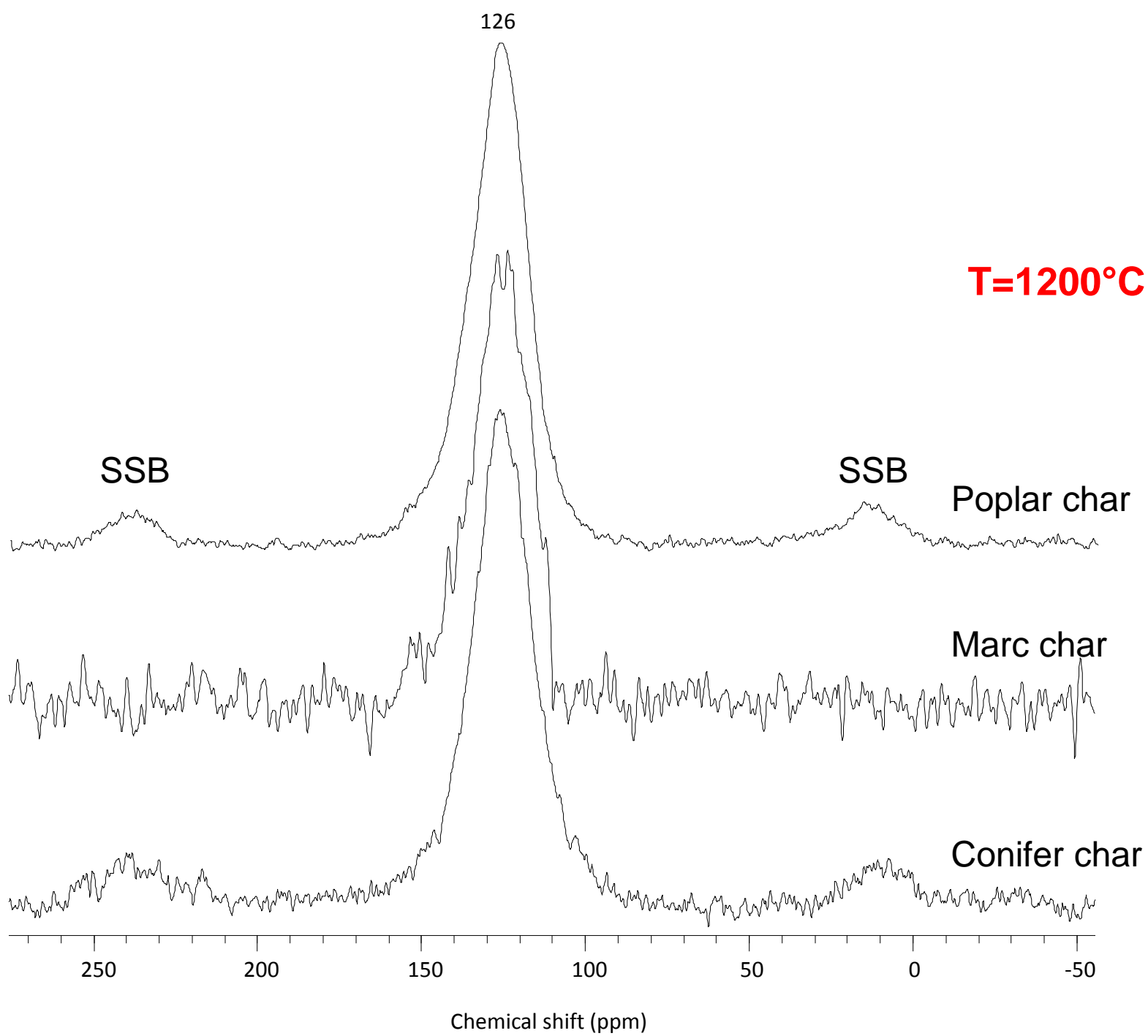
SOIL+POPLAR



SOIL+CONIFER

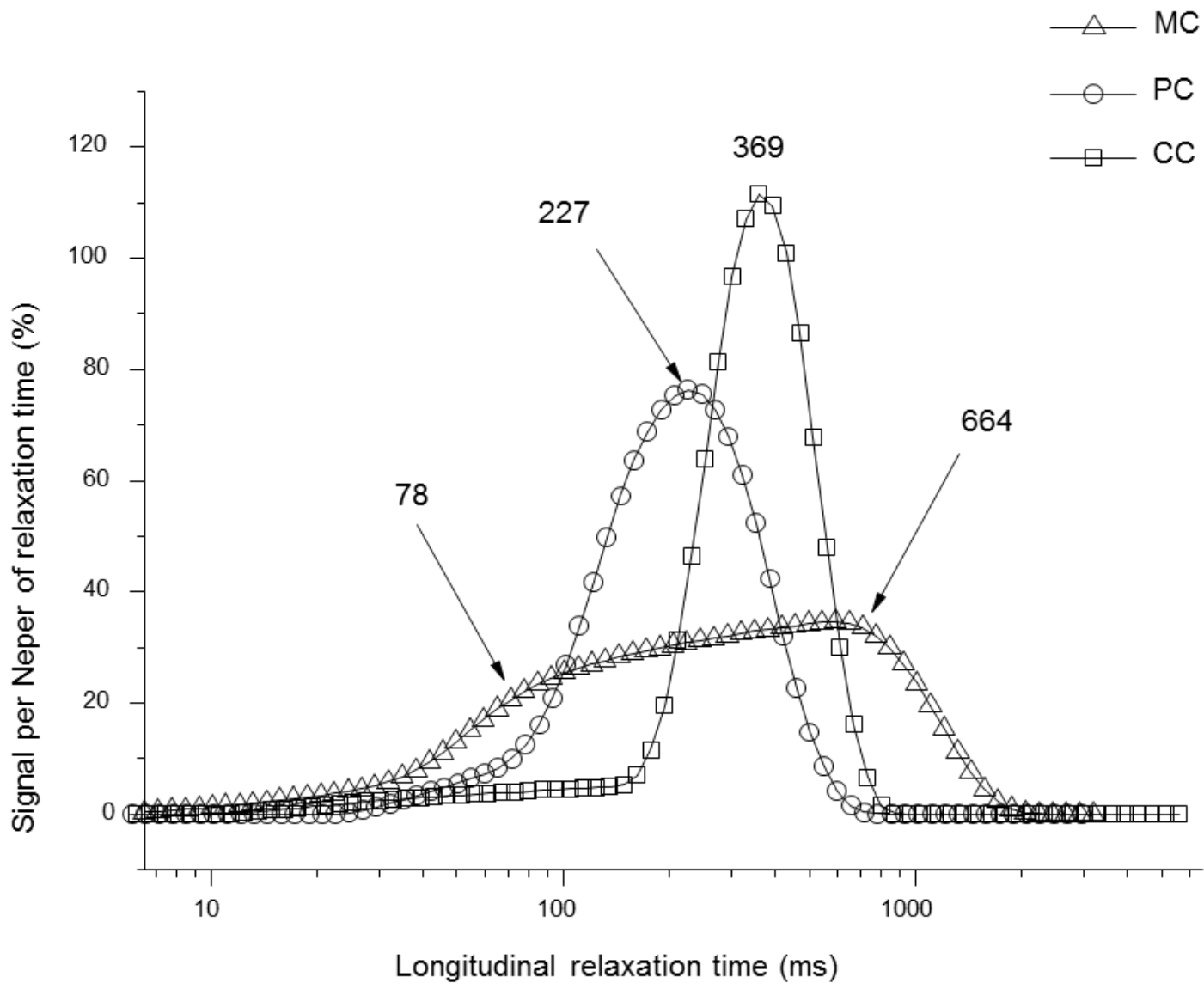


Samples	Surface area (m <sup>2</sup> ·g <sup>-1</sup> )	pH (in H <sub>2</sub> O)	Ashes	C*	N*	Na	K	Ca	Fe	Cu	Mn
Marc char	42±4	11.1±0.1	360±40	580±40	16±2	0.20±0.02	9.4±0.4	23.0±0.1	7.5±0.4	0.80±0.04	0.028±0.001
Poplar char	98±6	9.6±0.1	220±20	680±50	14±1	0.15±0.01	1.8±0.1	34.0±0.2	0.57±0.03	0.30±0.01	0.035±0.002
Conifer char	66±5	10.3±0.1	80±7	760±70	3.9±0.2	0.18±0.02	0.87±0.04	9.8±0.5	1.8±0.1	0.25±0.02	0.032±0.001



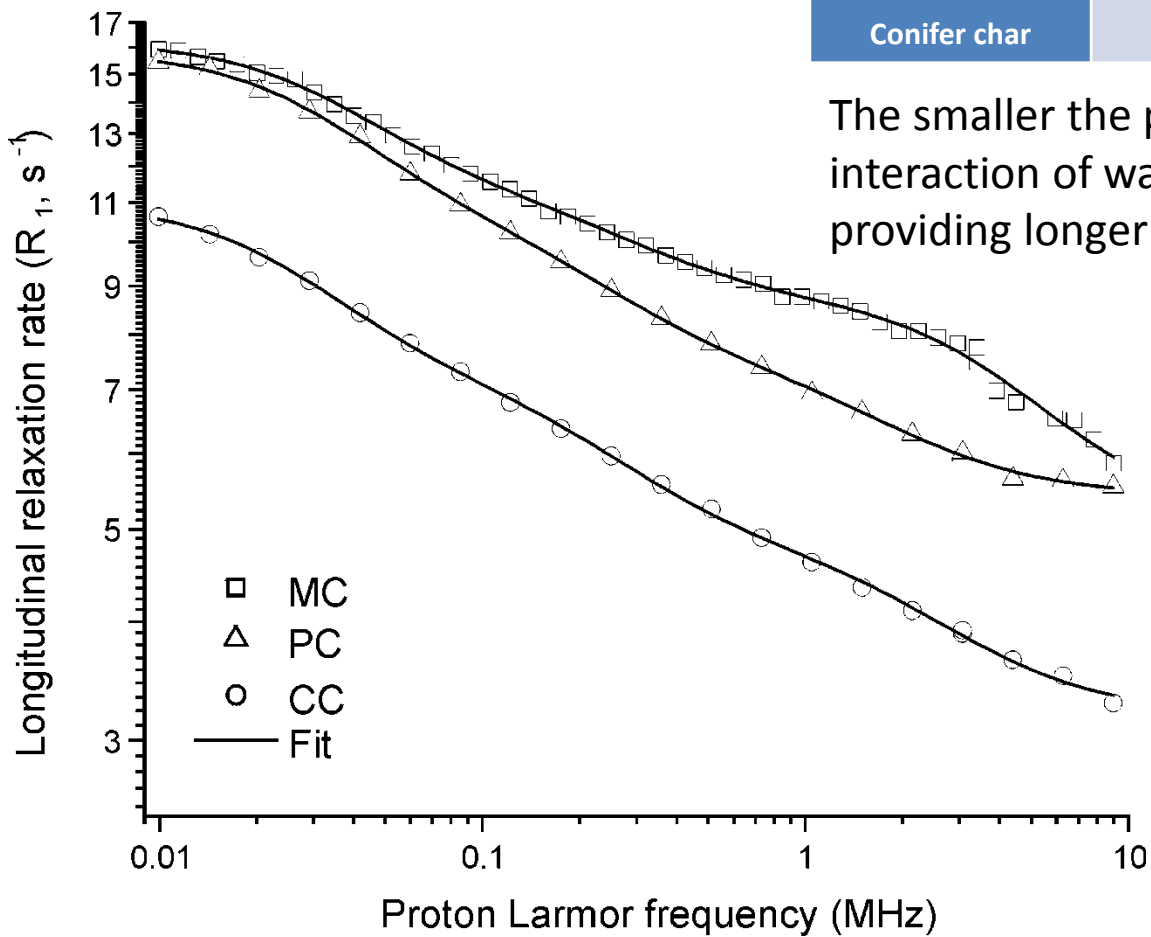
poplar char is richer in small sized pores, whereas large pore sizes appear to be characteristic for the conifer char.

Samples	Surface area ( $\text{m}^2 \cdot \text{g}^{-1}$ )
Marc char	42 $\pm$ 4
Poplar char	98 $\pm$ 6
Conifer char	66 $\pm$ 5



Samples	$\alpha$ (s <sup>-1</sup> )	$\beta$ (x 10 <sup>6</sup> s <sup>-2</sup> )	$\langle\tau\rangle$ (x 10 <sup>-6</sup> s <sup>-1</sup> )
Marc char	5.1±0.5	32±3	0.35±0.03
Poplar char	5.4±0.5	7.6±0.8	1.4±0.1
Conifer char	3.2±0.3	8.1±0.8	0.95±0.08

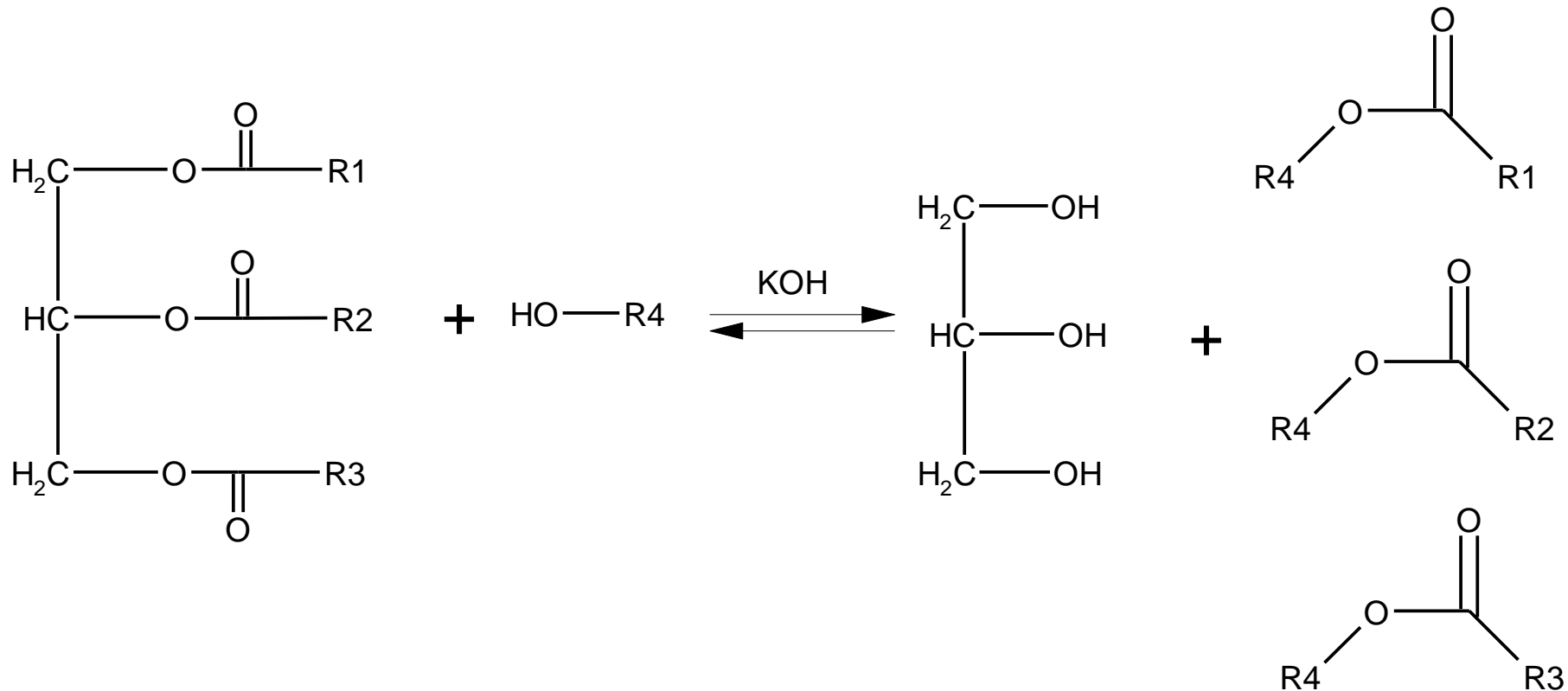
The smaller the pore size, the stronger is the interaction of water with the pore walls, thereby providing longer correlation time values



## **A case study**

# **Monitoring transesterification for biodiesel production**

# The reaction

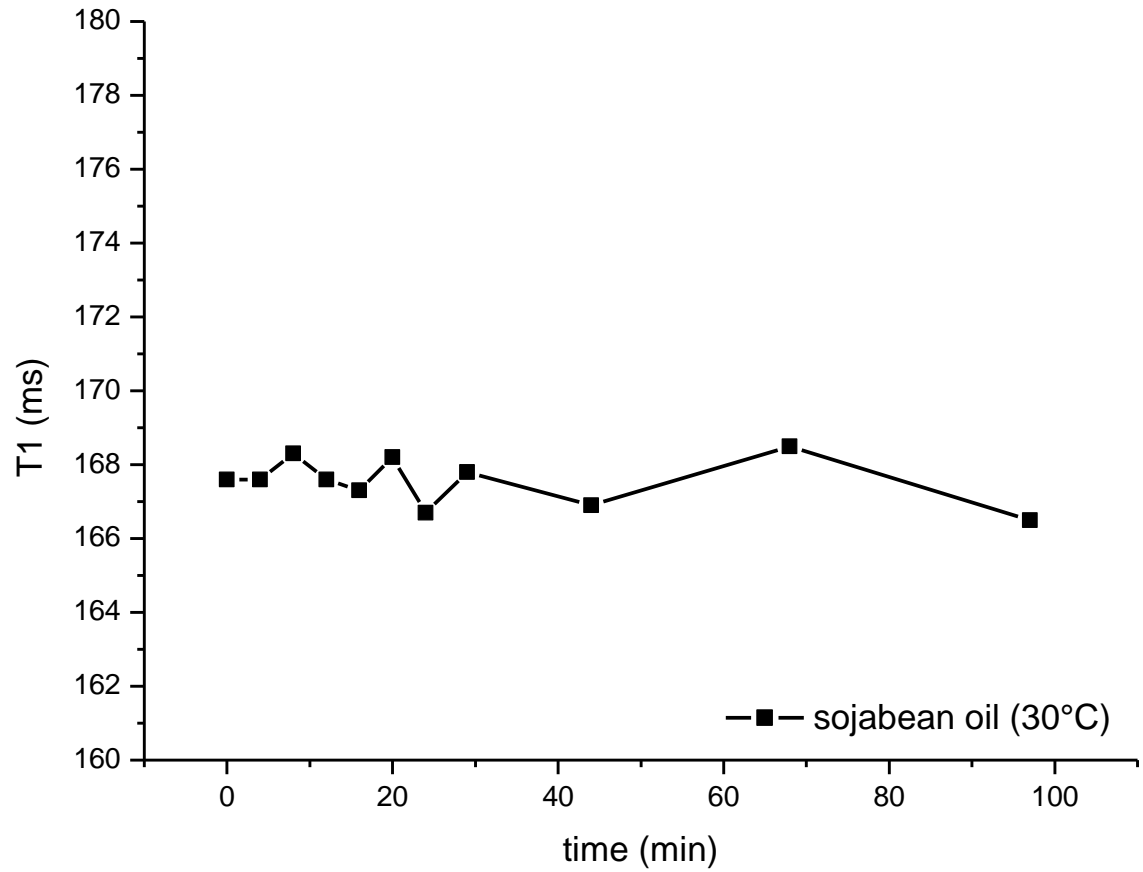




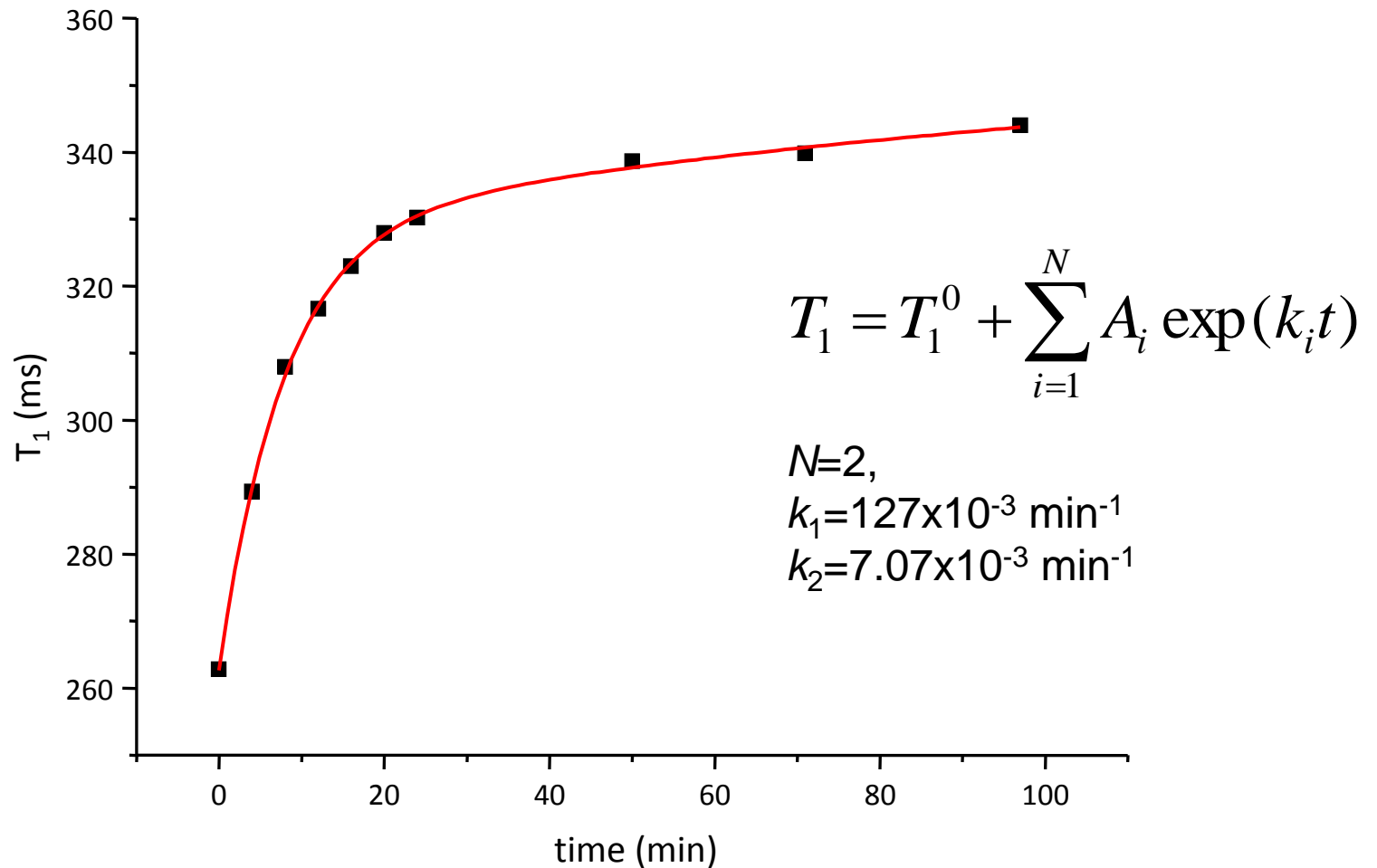
**Model used to account for the poly-exponential behavior of the recovery curves: stretched function**

$$y = y_0 + M \exp\left(-\frac{x}{T_1}\right)^k$$

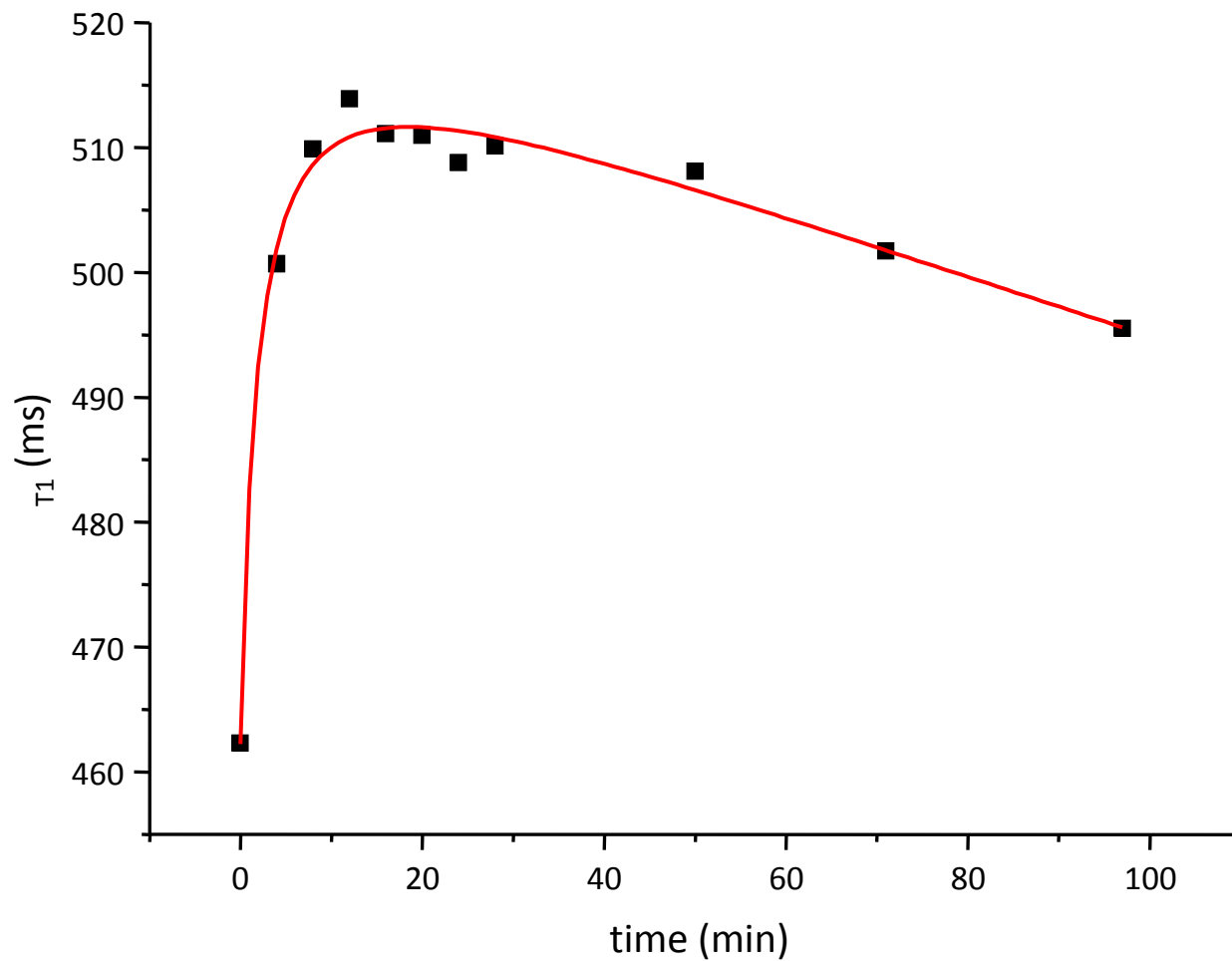
# Time $T_1$ variations of the sojabean oil at 30°C



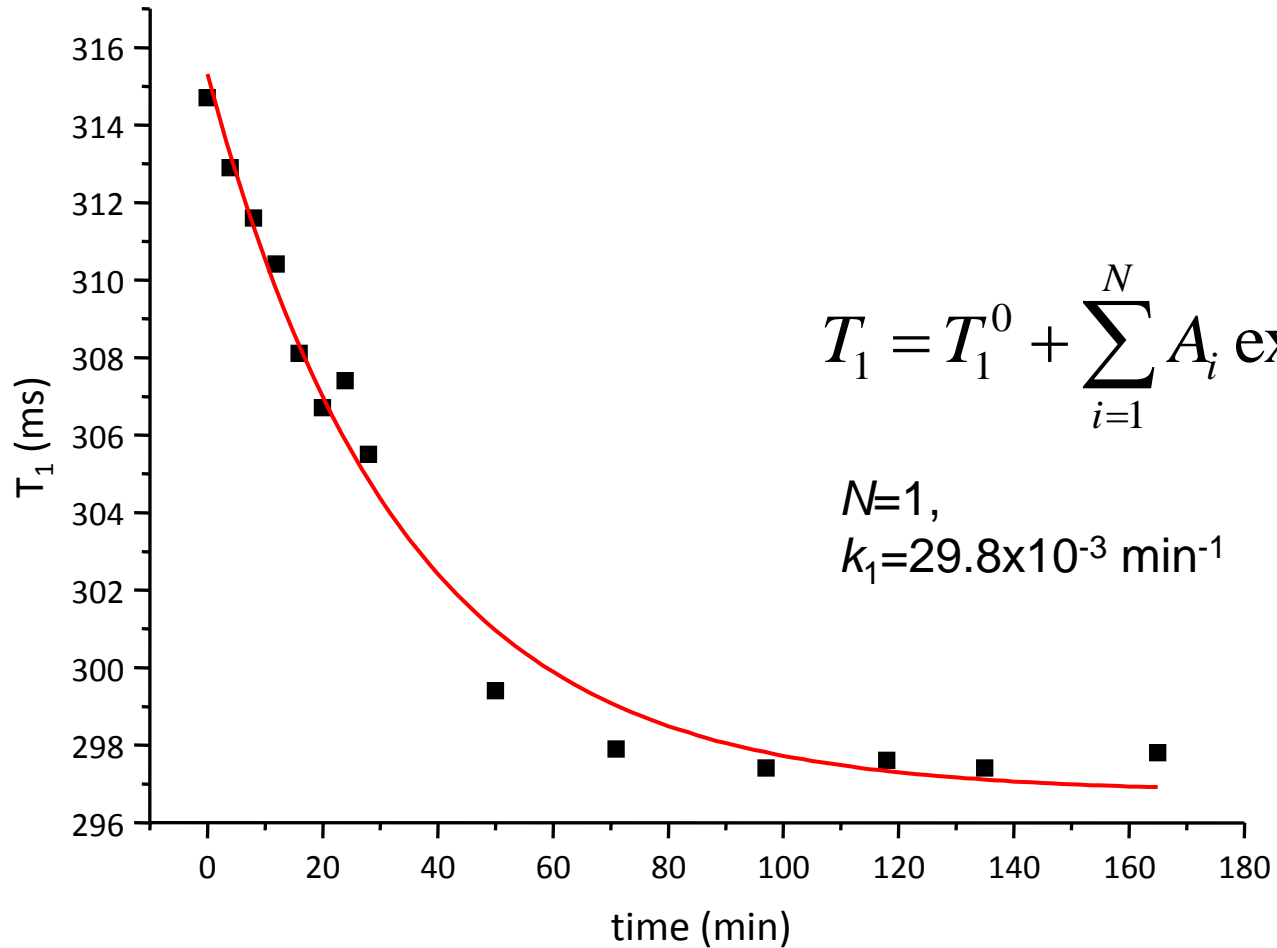
# Time $T_1$ variations of the sojabean oil added with MeOH/KOH (30°C)



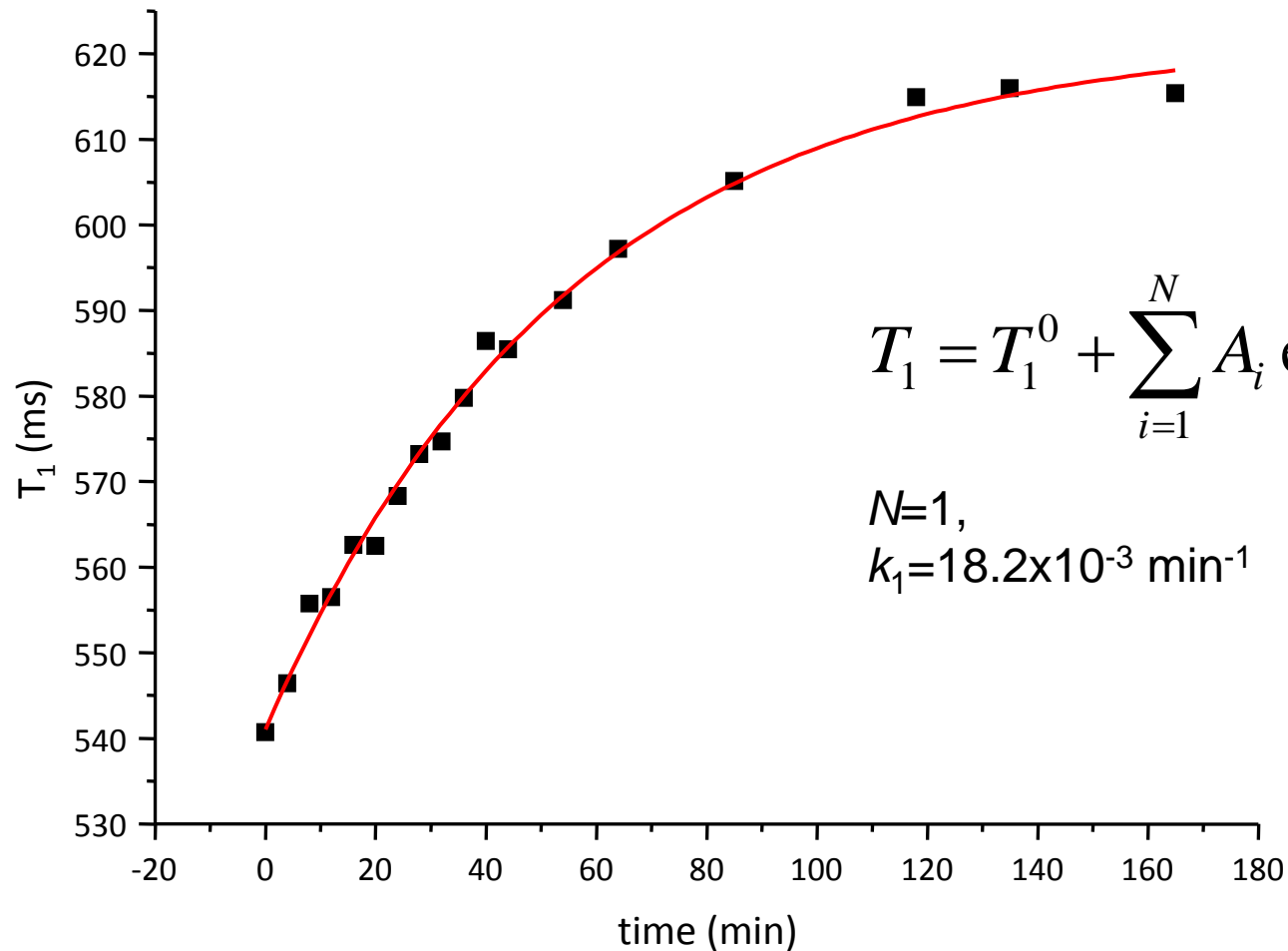
# Time $T_1$ variations of the sojabean oil added with EtOH/KOH (30°C)



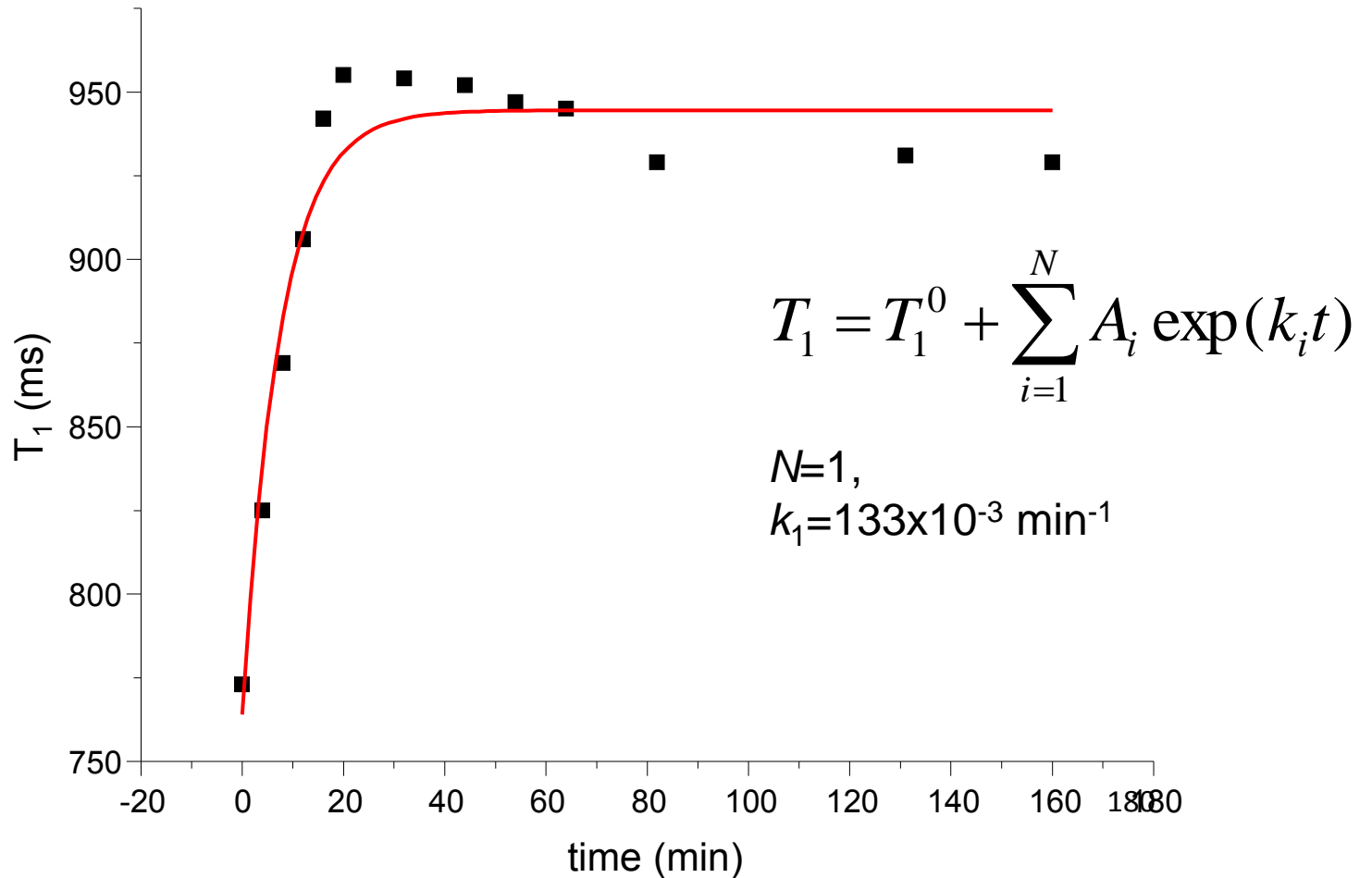
# Time $T_1$ variations of the sojabean oil added with BuOH/KOH (30°C)



# Time $T_1$ variations of the sojabean oil added with BuOH/KOH (50°C)



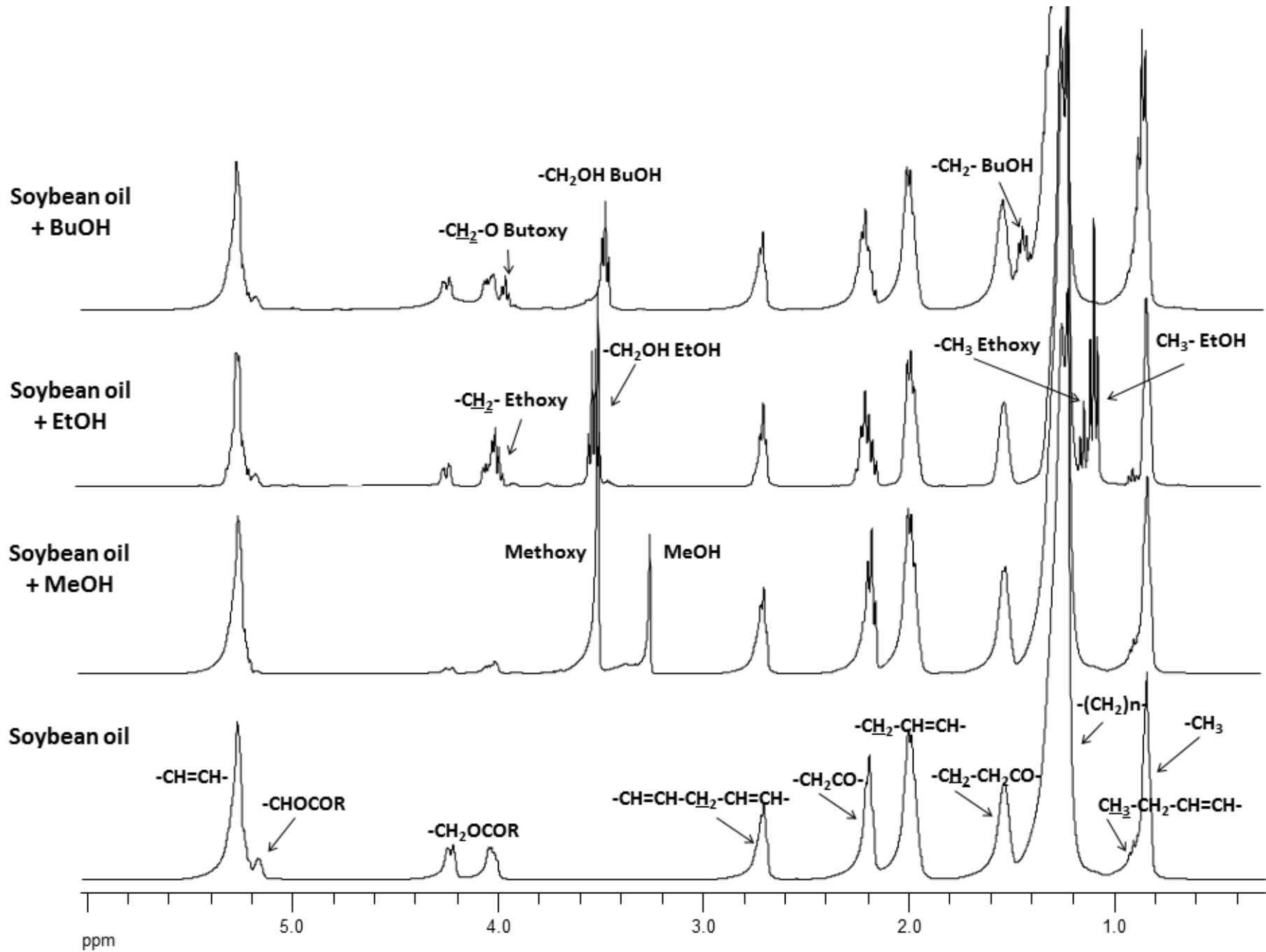
# Time $T_1$ variations of the sojabean oil added with BuOH/KOH (70°C)



# GC-MS analysis of the methyl-esters obtained at 30°C

KI	Area %	Compound	CAS
1680	0.50	Methyl tetradecanoate	000124-10-7
1779	0.07	Pentadecanoic acid, methyl ester	007132-64-1
1886	0.73	9-Hexadecenoic acid, methyl ester, (Z)-	001120-25-8
1938	21.85	Hexadecanoic acid, methyl ester	000112-39-0
2028	0.45	Heptadecanoic acid, methyl ester	001731-92-6
2093	28.61	8,11-Octadecadienoic acid, methyl ester	056599-58-7
2113	22.51	9,12-Octadecadienoic acid (Z,Z)-, methyl ester	000112-63-0
2175	2.70	9-Octadecenoic acid (Z)-, methyl ester	000112-62-9
2177	2.53	9,12,15-Octadecatrienoic acid, methyl ester, (Z,Z,Z)-	000301-00-8
2189	5.54	Octadecanoic acid, methyl ester	000112-61-8
2193	2.09	9,12-Octadecadienoic acid, methyl ester	002462-85-3
2196	0.40	10,13-Octadecadienoic acid, methyl ester	056554-62-2
2198	0.79	9,11-Octadecadienoic acid, methyl ester, (E,E)-	013038-47-6
2199	0.18	10-Nonadecenoic acid, methyl ester	056599-83-8
2292	0.70	11,13-Eicosadienoic acid, methyl ester	056599-57-6
2294	2.11	11-Eicosenoic acid, methyl ester	003946-08-5
2339	2.40	Eicosanoic acid, methyl ester	001120-28-1
2430	0.36	Heneicosanoic acid, methyl ester	006064-90-0
2531	3.09	Docosanoic acid, methyl ester	000929-77-1
2632	0.41	Tricosanoic acid, methyl ester	002433-97-8
2674	1.78	Tetracosanoic acid, methyl ester	002442-49-1
2872	0.20	Hexacosanoic acid, methyl ester	005802-82-4





**THANK YOU FOR YOUR ATTENTION**